

Cooling strategies for the atomization of glass-forming alloys

Nevaf Ciftci

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Zusammenfassung

Metallische Gläser bzw. massive metallische Gläser (bulk metallic glasses, BMG) sind relativ neue Werkstoffe, die aufgrund ihrer regellosen atomaren (amorphen) Anordnung gegenüber kristallinen Werkstoffen hohe Festigkeiten, Härten, elastische Dehnungen sowie gute Korrosionseigenschaften und besondere weichmagnetische Eigenschaften aufweisen. Außerdem zeichnen sie sich durch eine einmalige Prozessierbarkeit aus, da sie sich in einem bestimmten Temperaturfenster – ähnlich wie Kunststoffe – plastisch verformen lassen. Trotz ihrer relativ geringen plastischen Bruchdehnung werden sie häufig als Strukturwerkstoffe in Erwägung gezogen und haben bereits erste kristalline Werkstoffe in industriellen Anwendungen (Transformatoren, Spulen, Induktoren, Federn, Gehäuse, Skalpelle, Sportartikel etc.) ersetzt.

Die Herstellbarkeit von amorphen Produkten ist abhängig von der Legierungszusammensetzung und den Abkühlbedingungen. Die Legierung muss schnell abgekühlt werden, damit sich die Atome nicht zu einer Kristallstruktur ordnen können. Die Schmelze erstarrt bei der Glasübergangstemperatur kristallin, wenn die Abkühlrate nicht ausreichend ist. Die erforderlichen Abkühlraten führen dazu, dass die erzielbaren Bauteilabmessungen auf kleine Dimensionen beschränkt sind. Diese geometrischen Restriktionen können durch einen zweistufigen Prozess aus Pulververdüsung und anschließender Pulververdichtung überwunden werden. Zerstäubungstechniken sind vielversprechende Techniken, da die Schmelzetropfen durch das große Oberflächen-Volumen-Verhältnis in den amorphen Zustand überführt werden können. Bei der Pulververdüsung werden typischerweise Abkühlraten von 10^2 bis 10^5 K s^{-1} erreicht.

Das Ziel der Dissertation war die Erweiterung der Abkühlstrategien bei der Zerstäubung glasbildender Schmelzen. Der Fokus lag auf der Verdüsung von Legierungen mit Einschmelzmaterialien, die zu moderaten Preisen erhältlich sind. Diese Glasbilder können aufgrund des geringeren Reinheitsgrades (commercial purity, CP) im Vergleich zu Materialien mit Laborreinheit (high purity, HP) nur durch höhere Abkühlraten amorph verdüst werden. Hierbei war es notwendig, den Wärmeübergangskoeffizienten zu erhöhen. Dies setzte eine gezielte Abstimmung hinsichtlich der Prozessgrößen voraus, sodass neben Legierungen mit geringen Reinheitsgraden auch größere Schmelzetropfen amorph verdüst werden konnten. Durch den Einsatz verschiedener Abkühlstrategien sollten neue Prozessfenster eröffnet werden, die bei der konventionellen Pulververdüsung nicht erreicht werden. Die entwickelten Abkühlstrategien wurden erstmalig auf glasbildende Legierungen übertragen: (i) Variation der Überhitzungstemperatur, (ii) Heißgaszerstäubung, (iii) Sprühkegelkühlung sowie (iv) Flüssigabschreckung. Diese Abkühlstrategien hatten zusätzlich das Ziel, heterogene Keimbildung zu minimieren und kleinere Partikel zu erzeugen, da die Abkühlrate der Schmelzetropfen während der Erstarrungsphase im Wesentlichen von der Partikelgröße und dem GMR (Massenstromverhältnis aus Gas zu Metallschmelze) abhängig ist.

Die höchsten Abkühlraten wurden für flüssig-abgeschreckte Partikel erreicht. Die $\{(Fe_{0,6}Co_{0,4})_{0,75}B_{0,2}Si_{0,05}\}_{96}Nb_4$ bzw. $Fe_{76}B_{10}Si_9P_5$ Pulver konnten dabei vollständig amorph erzeugt werden ($< 200 \mu m$), wohingegen FeCoBSiNb Partikel bei der konventionellen Pulververdüsung nur bis zu $63 \mu m$ amorph waren. Selbst die kleinsten FeBSiP Partikel ($< 25 \mu m$) waren bei der konventionellen Pulververdüsung nur teilmorph. Ferner zeigte sich, dass induzierter Wasserstoff ($10^2 - 10^4$ ppm) beim Abschrecken mit Wasser die Glasbildungseigenschaften und die Sättigungsmagnetisierung der Pulver verbesserte. Eine weitere Fragestellung im Rahmen der Arbeit war, inwieweit die flüssig-abgeschreckten Partikel weiterverarbeitet werden können, da ein starker Oxidfilm um die Partikel vermutet wurde. Mittels Spark Plasma Sintering konnten unabhängig von den verwendeten Abkühlstrategien amorphe Ringe mit einer relativen Dichte $> 98\%$ erzeugt werden. Die Ringe zeigten dabei eine nahezu unveränderte Sättigungsmagnetisierung, wobei die Koerzitivfeldstärke leicht anstieg bzw. die Permeabilität stark abnahm. Durch den Einsatz der Flüssigabschreckung konnten diese Legierungen mit CP erstmalig in feste Probekörper gepresst werden und bieten dabei durch ihre guten magnetischen Eigenschaften (unabhängig von der Reinheit) ein hohes Potential für industrielle Anwendungen.

Die zweithöchsten Abkühlraten wurden für die Abkühlstrategie Sprühkegelkühlen beobachtet. Die FeCoBSiNb bzw. FeBSiP Partikel waren bis zu $125 \mu m$ bzw. $45 \mu m$ vollständig amorph. Eine erhöhte Überhitzungstemperatur führte trotz der Annahme, dass Nukleationskeime in der Schmelze aufgelöst bzw. verringert werden nicht zu einer signifikanten Erhöhung des amorphen Anteils. Es zeigte sich jedoch, dass höhere Überhitzungstemperaturen die Partikelformfaktoren verbesserten. Höhere Formfaktoren führen dabei zu einer höheren Pulverfließfähigkeit. Es wurden neben den beiden Fe-Legierungen zwei weitere Legierungen betrachtet: $Zr_{59,3}Cu_{28,8}Al_{10,4}Nb_{1,5}$ und $Cu_{47}Ti_{33}Zr_{11}Ni_6Si_1Sn_2$. Die ZrCuAlNb Legierung wies unabhängig von der Überhitzungstemperatur eine hohe Anzahl an nicht-sphärischen Partikeln auf. Dieses Verhalten war konträr zu den anderen Legierungen und kann auf unterschiedliche Viskositäten zurückgeführt werden. Eine hohe Schmelzeviskosität beeinflusst maßgeblich die Tropfenverformung.

Erhöhte Zerstäubergastemperaturen führten zu einer geringeren Gasdichte und damit zu einer höheren Gas- bzw. Relativgeschwindigkeit, wobei der Temperaturgradient zwischen Schmelzetropfen und Gas abnahm. Dieser gegenseitige Effekt führte zu einer Kompensation, sodass der amorphe Anteil unverändert blieb bzw. leicht abnahm. Neue semi-empirische Korrelationen zur Bestimmung der Abkühlraten bestätigten dabei die experimentellen Beobachtungen. Die Heißgaszerstäubung ist jedoch eine vielversprechende Methode zur industriellen Großproduktion von amorphen Metallpulvern. Am Beispiel der FeCoBSiNb Legierung wurde gezeigt, dass durch die verringerte Gasdichte des Zerstäubergases signifikant kleinere Partikel erzeugt wurden und der Gasverbrauch um bis zu 50% gesenkt werden konnte. Ferner konnte die amorphe Pulverausbringung um ca. 20% erhöht werden. Jedoch verschlechterten sich die Partikelformfaktoren zu geringeren Werten.

Summary

Metallic glasses or bulk metallic glasses are a new class of materials. These amorphous metals have a disordered structure, resulting in outstanding properties when compared to crystalline materials such as high hardness, strength, elasticity, good wear and corrosion resistance, and superior soft magnetic properties. Similar to thermoplastics, they are easily deformable due to a highly viscous state between the glass transition temperature and the crystallization temperature. Despite their low plasticity and tendency for catastrophic failure through instantaneous shear band propagation, they are often considered as structural materials and have already been implemented in industrial applications (transformers, coils, inductors, springs, micro gears, cases, scalpels, sporting goods etc.).

The properties of metallic glasses depend on alloy composition and cooling conditions during solidification. The liquid must be cooled rapidly to avoid crystallization, forcing the melt to retain its liquid atomic structure. The required cooling rates to obtain an amorphous solid are limited by time-dependent heat conduction, restricting product dimensions. Geometric limitations can be overcome by powder synthesis and consolidation to introduce metallic glasses to the commercial market. Gas atomization is a promising technique in the commercial production of metallic glasses, as amorphous particles can be produced due to the high surface-to-volume ratio through conduction and radiation. The cooling rate in molten gas atomization typically ranges from 10^2 to 10^5 K s^{-1} .

The aim of this PhD thesis was the development of novel cooling strategies for the atomization of glass-forming alloys to increase cooling rates during droplet solidification. The focus was on the atomization of soft ferromagnetic glass-forming alloys with commercial purity. These alloys are difficult to atomize into an amorphous state due to cooling rate limitations and their low glass-forming ability. For this purpose, it was necessary to increase the heat transfer coefficient. This resulted in higher cooling rates, allowing for the atomization of soft ferromagnetic glass-forming alloys with commercial purity and larger particle sizes that normally tend to crystallize during droplet solidification. With the development of novel cooling strategies, new process windows have been made available that are typically inaccessible in conventional gas atomization. Therefore, four cooling strategies were developed for the atomization of glass-forming alloys: (i) increasing the melt superheat temperature, (ii) hot gas atomization, (iii) spray cone cooling, and (iv) liquid quenching. The developed cooling strategies were used to decrease potent nucleation sites in the melt as well as to produce smaller particles, as the cooling rate strongly depends on droplet size and gas-to-melt mass flow ratio.

The highest cooling rates were achieved for liquid quenched particles, as the $\{(\text{Fe}_{0.6}\text{Co}_{0.4})_{0.75}\text{B}_{0.2}\text{Si}_{0.05}\}_{96}\text{Nb}_4$ and $\text{Fe}_{76}\text{B}_{10}\text{Si}_9\text{P}_5$ powders were fully amorphous for the entire particle size range (< 200 μm). Conventionally atomized FeCoB -

SiNb particles were only fully amorphous up to 63 μm and FeBSiP particles were partially-crystalline even for the smallest particle size class ($< 25 \mu\text{m}$). Hydrogen inducement during water quenching ($10^2 - 10^4$ ppm) had a positive effect on the glass-forming ability and saturation magnetization. Since it was expected that an oxide film would be formed during liquid water quenching, the powder consolidation ability was also investigated. Amorphous rings with a relative density $> 98\%$ were produced via Spark Plasma Sintering, regardless of the cooling strategies used. The saturation magnetization of the sintered rings remained nearly constant, while the coercivity slightly increased and the permeability distinctly decreased. Liquid quenching resulted in an increased amorphous fraction and improved powder properties as well as consolidation feasibility, making liquid quenching an influential opportunity for the commercialization of these alloys with commercial purity feedstock.

The second highest cooling rates were achieved for spray cone cooling. The FeCoBSiNb and FeBSiP particles were amorphous up to 125 μm and 45 μm , respectively. An increased melt superheat temperature resulted in a slightly higher amorphous fraction. It was assumed that a high melt superheat temperature reduced or destroyed any potential nucleation sites in the melt.

An increased melt superheat temperature ranging from 250 to 450 K significantly increased particle shape factors such as sphericity and aspect ratio. For the particle shape study, a $\text{Zr}_{59.3}\text{Cu}_{28.8}\text{Al}_{10.4}\text{Nb}_{1.5}$ and a $\text{Cu}_{47}\text{Ti}_{33}\text{Zr}_{11}\text{Ni}_6\text{Si}_1\text{Sn}_2$ (at%) alloy were also selected. The ZrCuAlNb glass-forming alloy showed a significantly higher amount of non-spherical particles and had a rougher surface, while the three other glass-forming alloys exhibited smooth surfaces and a nearly spherical particle morphology. This is due to the higher viscosity as the droplet break-up and formation process depends on viscosity.

The FeCoBSiNb glass-forming alloy was used to demonstrate the effect of hot gas atomization on cooling rates. A higher gas velocity was introduced by a decreased gas density (through elevated gas temperatures), leading to a higher relative velocity, while the temperature gradient between the melt droplet and surrounding gas was reduced. The relative velocity and the temperature gradient compensated for each other and resulted in nearly unchanged cooling rates. Thus, the amorphous fraction remained the same or slightly decreased. Newly developed semi-empirical correlations to predict average cooling rates confirmed these results. Despite the fact that the cooling rate remained constant, hot gas atomization provided many advantages over conventional gas atomization. For instance, hot gas atomization is a promising design tool for the commercialization of bulk metallic glasses with the benefits of producing smaller amorphous powders, reducing gas consumption up to 50%, retaining the amorphous fraction, and increasing the amorphous yield by 20%. By contrast, particle shape factors decreased due to the attachment of smaller particles on larger semi-solidified droplets known as satellites. An increase in particle shape factors typically results in improved powder flowability.

List of publications and conference contributions

List of publications

[1] N. Ciftci, N. Ellendt, L. Mädler, V. Uhlenwinkel. Impact of Hot Gas Atomization on Glass Forming Alloys. Hamburg: World PM 2016, European Powder Metallurgy Association, ISBN: 978-1-899072-48-4, 2016.

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(110206):1-9, 2020.

Conference contributions

[1] N. Ciftci, N. Ellendt, L. Mädler, V. Uhlenwinkel. Impact of Hot Gas Atomization on Glass Forming Alloys. World PM 2016 Congress & Exhibition, European Powder Metallurgy Association, Hamburg, Germany, 09-13 October 2016.

[2] N. Ciftci, N. Ellendt, L. Mädler, V. Uhlenwinkel. Cooling strategies for the droplet solidification of glass-forming alloys. 25th International Conference on Metastable, Amorphous, and Nanostructured Materials (ISMANAM 2018), Rome, Italy, 02-06 July 2018.

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